# organic papers

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#### **Key indicators**

Single-crystal X-ray study T = 150 K Mean  $\sigma$ (C–C) = 0.003 Å R factor = 0.031 wR factor = 0.075 Data-to-parameter ratio = 8.5

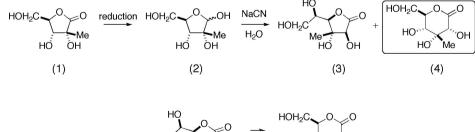
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

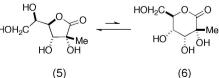
# 3-C-Methyl-D-allono-1,5-lactone

The relative configuration and ring size of the title compound,  $C_7H_{12}O_6$ , were established by X-ray crystallographic analysis. The absolute configuration was determined by the use of 2-*C*-methyl-D-ribonolactone as a starting material. Almost all unprotected carbohydrate lactones are five-membered ring 1,4-lactones; the title compound provides a very rare example of a stable six-membered ring lactone.

### Comment

Although carbohydrates are the most varied of cheap chiral building blocks (Lichtenthaler & Peters, 2004), only recently have the first examples of branched 2-C-methylpentoses become readily available by treatment of an Amadori ketose with aqueous calcium hydroxide (Hotchkiss et al., 2007). The recognition of the value of a family of 2-C-methylnucleosides in the treatment of hepatitis C has led to current interest in the synthesis of 2-C-carbon-substituted sugars (Sorbera et al., 2006). The Kiliani reaction of ketoses and deoxyketoses with cyanide has provided an environmentally friendly procedure for the generation of a set of carbohydrate scaffolds with a branched carbon substituent at C-2 (Hotchkiss et al., 2004; Soengas et al., 2005). X-ray crystallographic analysis was vital in determining the structures of the products in these reactions (Punzo et al., 2006; Watkin et al., 2005). At present, free sugars and their lactones with a carbon branch at C-3 are essentially unknown. A 3-C-methylpentonolactone of unknown stereochemistry has been isolated from cigarette smoke (Schumacher et al., 1977), 3-C-methyl-D-mannose (Kwon et al., 2004) is one of the components of the trisaccharide repeating unit of the polysaccharide from Helicobacter pylori (Kocharova et al., 2000) and 3-C-methyl-L-mannose is one of the sugars in a pentasaccharide hapten of the GPL of Mycobacterium avium serovar (Fekete et al., 2006).

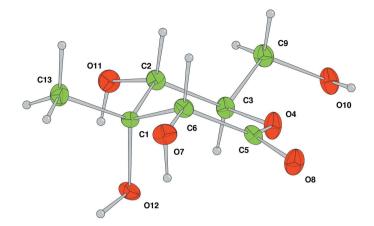




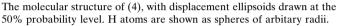
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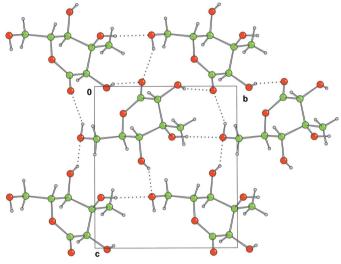
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#### Figure 1





#### Figure 2

A projection along the a axis of part of a hydrogen-bonded sheet of (I). There are no strong interactions between the sheets. Hydrogen bonds are shown as dotted lines.

The value of the Kiliani reaction on 2-C-carbon-substituted carbohydrates in the synthesis of 3-C-hydroxymethyl branched sugars (Parker et al., 2006; Simone et al., 2007) and 3-Cmethyl branched sugars (Bream et al., 2006) has been established. Under completely environmentally friendly aqueous conditions, the reaction of cyanide in water with 2-C-methyl-D-ribose, (2), derived from 2-C-methyl-D-ribonolactone, (1) (Hotchkiss et al., 2006), gave a major product which crystallized from the reaction mixture. X-ray crystallographic analysis shows (Fig. 1) that the structure is the title compound, (4), removing ambiguities as to the stereochemistry at the new C-2 chiral centre and the ring size of the lactone. The minor product is most likely the five-membered ring altrono-lactone, (3). The strain in five-membered ring lactones is generally considerably less than in six-membered ring lactones (Luisa et al., 1990; Brown et al., 1989). Compound (4) is thus a very rare example of the preferential formation of a six-membered ring lactone. Its C-2 isomer crystallizes as the 2-C-methyl-D-allono1,4-lactone, (5), rather than the six-membered ring isomer, (6) (Harding *et al.*, 2005). The absolute configuration of compound (4) was determined by the use of the D-sugar (1) as the starting material.

The isolated molecule of (4) (Fig. 1) shows no unusual bond lengths or angles, in spite of the strain mentioned above. The largest differences from the *MOGUL* norms (Bruno *et al.*, 2004) are C6–O7 (00.01 Å; *MOGUL* s.u. 0.02 Å) and C3–C5–O4 (5.0°; *MOGUL* s.u. 2.2°).

The crystal structure of (4) is composed of hydrogenbonded sheets of molecules lying parallel to the bc plane. Both the ketonic atom O8 and the hydroxyl atom O10 act as acceptors for two hydrogen bonds (Table 1 and Fig. 2).

## **Experimental**

3-*C*-Methyl-D-allono-1,5-lactone, (4), was crystallized from a 3:1:1 mixture of ethyl acetate, methanol and cyclohexane. Analysis: m.p. 421–423 K;  $[\alpha]_D^{23}$  63.5 (*c*, 0.795 in MeOH).

Z = 2

 $D_x = 1.551 \text{ Mg m}^{-3}$ 

Mo  $K\alpha$  radiation

Needle, colourless

 $0.20 \times 0.05 \times 0.05 \text{ mm}$ 

5511 measured reflections

1000 independent reflections

930 reflections with  $I > 2\sigma(I)$ 

 $\mu = 0.14 \text{ mm}^{-1}$ 

T = 150 K

 $R_{\rm int}=0.050$ 

 $\theta_{\rm max} = 27.4^{\circ}$ 



 $C_7H_{12}O_6$   $M_r = 192.17$ Monoclinic,  $P2_1$  a = 5.6603 (2) Å b = 8.0045 (2) Å c = 9.3242 (3) Å  $\beta = 103.1470 (13)^{\circ}$   $V = 411.39 (2) Å^3$ 

#### Data collection

Nonius KappaCCD area-detector diffractometer ω scans

Absorption correction: multi-scan (*DENZO* and *SCALEPACK*; Otwinowski & Minor, 1997)  $T_{min} = 0.75, T_{max} = 1.0$ 

### Refinement

$w = 1/[\sigma^2(F^2) + (0.02P)^2]$
+ 0.12P]
where $P = [\max(F_o^2, 0) + 2F_c^2]/3$
$(\Delta/\sigma)_{\rm max} < 0.001$
$\Delta \rho_{\rm max} = 0.29 \ {\rm e} \ {\rm \AA}^{-3}$
$\Delta \rho_{\rm min} = -0.26 \text{ e } \text{\AA}^{-3}$

Table 1		
Hydrogen-bond geometry	(Å.	°)

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
O10−H7···O8 <sup>i</sup>	0.84	2.08	2.856 (2)	153
$O7-H8\cdots O8^{ii}$	0.85	2.06	2.832 (2)	152
$O11 - H12 \cdots O10^{iii}$	0.82	2.09	2.836 (2)	152
O12−H3···O10 <sup>iv</sup>	0.83	2.07	2.901 (2)	173

Symmetry codes: (i) -x + 2,  $y - \frac{1}{2}$ , -z; (ii) -x + 2,  $y + \frac{1}{2}$ , -z; (iii) -x + 2,  $y + \frac{1}{2}$ , -z + 1; (iv) x, y + 1, z.

In the absence of significant anomalous scattering, Friedel pairs were merged and the absolute configuration assigned from the starting material. The relatively large ratio of minimum to maximum corrections applied in the multiscan process (1:1.33) reflects changes in the illuminated volume of the very thin needle-like crystal. These changes were kept to a minimum and were taken into account (Görbitz, 1999) by multiscan interframe scaling (*DENZO* and *SCALEPACK*; Otwinowski & Minor, 1997).

The H atoms were all located in a difference map, but those attached to C atoms were repositioned geometrically. The H atoms were initially refined with soft restraints on the bond lengths and angles to regularize their geometry, with C-H in the range 0.93–0.98 Å and O-H = 0.82 Å, and with  $U_{\rm iso}({\rm H}) = 1.2-1.5U_{\rm eq}$ (parent atom), after which the positions were refined with riding constraints.

Data collection: *COLLECT* (Nonius, 1997–2001); cell refinement: *DENZO* and *SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *DENZO* and *SCALEPACK*; program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *CRYSTALS* (Betteridge *et al.*, 2003); molecular graphics: *CAMERON* (Watkin *et al.*, 1996); software used to prepare material for publication: *CRYSTALS*.

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